

Tetrahedron Letters 46 (2005) 3291-3295

Tetrahedron Letters

## Chemo-enzymatic short-step total synthesis of symbioramide

## Tsukasa Takanami, Hiroki Tokoro, Dai-ichiro Kato, Shigeru Nishiyama and Takeshi Sugai\*

Department of Chemistry, Keio University, 3-14-1 Hiyoshi, Yokohama 223-8522, Japan Received 28 February 2005; revised 11 March 2005; accepted 16 March 2005 Available online 1 April 2005

Abstract—A concise synthesis of symbioramide, a marine-origin ceramide from a common starting material, methyl (±)-trans-2,3epoxyoctadecanoate, in a convergent manner was achieved. The key step is the direct lipase-catalyzed coupling reaction between methyl (2R,3E)-2-hydroxy-3-octadecenoate and non-protected (±)-erythro-dihydrosphingosine, giving natural (2S,3R,2'R)-symbioramide and its (2R,3S,2'R)-isomer in 38% and 37% yield, respectively. The optically active  $\beta$ , $\gamma$ -unsaturated  $\alpha$ -hydroxyester was prepared by Mg(ClO<sub>4</sub>)<sub>2</sub>-mediated isomerization of epoxide and the subsequent lipase PS-catalyzed kinetic resolution. © 2005 Elsevier Ltd. All rights reserved.

Symbioramide (1a), a naturally occurring marine ceramide as a Ca<sup>2+</sup>-ATPase activating factor as well as the antileukemic substance against L1210 murine leukemia cells, consists of the D-erythro-dihydrosphingosine [(2S,3R)-2] and  $\beta,\gamma$ -unsaturated  $\alpha$ -hydroxycarboxylic acid (2R,3E)-3a. Three groups have been successful in chemical synthesis of **1a** and related stereoisomers.<sup>2–4</sup>

The present synthetic strategy has the following feature. Both the dihydrosphingosine and the long-chain  $\beta,\gamma$ -unsaturated  $\alpha$ -hydroxycarboxylic acid moiety are derived from the common intermediate, an  $\alpha,\beta$ -epoxycarboxylic ester (4). Toward this end, the isomerization of the  $\alpha,\beta$ -epoxy ester ( $\pm$ )- $4^5$  into  $\beta,\gamma$ -unsaturated  $\alpha$ hydroxyester 5a is the first task (Fig. 1).

So far, some perchlorates<sup>6</sup> had been reported for similar transformations, and indeed, LiClO<sub>4</sub> promoted the desired reaction (Table 1). Among Lewis acids, whose counteranions have low nucleophilicity, Mg(ClO<sub>4</sub>)<sub>2</sub><sup>7</sup> yielded the best result (61%). The reaction was stopped within 2 h at 110 °C in toluene, 8 to avoid the further decomposition of 5a under prolonged incubation. The product 5a, obtained as an inseparable mixture of isomers (E/Z = 4:1, Scheme 1), was once oxidized to the  $\alpha,\beta$ -unsaturated ketone [(E)- $\mathbf{6}$ , 994%] with Dess–Martin

Entry	Lewis acid	Temp (°C)	Yield (%)
1	LiClO <sub>4</sub>	110	32
2	$Zn(ClO_4)_2$	110	23
3	$In(ClO_4)_3$	70	29
4	$Mg(ClO_4)_2$	110	61
5	LiOTf	110	No reaction
6	$Mg(OTf)_2$	110	No reaction
7	$Zn(OTf)_2$	110	21

All of the reaction were performed in toluene.

symbioramide

(2S,3R,2'R)-1a

Keywords: Symbioramide; Dihydrosphingosine; α-Hydroxy acid; Lipase; Aminolysis.

reagent. Then, the ketoester was reduced to afford the original hydroxyester 5a with a variety of reducing reagents. In most cases, however, some unexpected basic conditions caused re-isomerization of double bond. The use of L-Selectride<sup>10</sup> in Et<sub>2</sub>O/hexane (10:1) at

Figure 1. **Table 1.** Isomerization of epoxy ester to  $\beta, \gamma$ -unsaturated  $\alpha$ -hydroxy-

<sup>\*</sup> Corresponding author. Tel.: +81 45 566 1709; fax: +81 45 566 1697; e-mail: sugai@chem.keio.ac.jp

OH  

$$C_{14}H_{29}$$
 $C_{02}Me$ 
 $C_{14}H_{29}$ 
 $C_{02}Me$ 
 $C_$ 

Scheme 1. Reagents and conditions: (a) Mg(ClO<sub>4</sub>)<sub>2</sub>, MS3A, toluene, 110 °C, 61%; (b) Dess–Martin periodinane, CH<sub>2</sub>Cl<sub>2</sub>, 94%; (c) L-Selectride, Et<sub>2</sub>O/hexane (10:1), -78 °C, 75%; (d) Chirazyme L-2, phosphate buffer (0.1 M, pH 7.0), 30 °C, quant.

-78 °C, gave the highest ratio (E/Z = 22:1) of the product (75%). The ester was hydrolyzed (quant.) under as mild as neutral conditions by the treatment with *Candida antarctica* lipase (Chirazyme L-2) in buffer, and the recrystallization of the product provided pure (2RS,3E)-3a (Scheme 1). 12

The enantiomeric resolution was successful by applying vinyl acetate and *Pseudomonas cepacia* lipase (Amano PS).<sup>13</sup> The subsequent  $CH_2N_2$  treatment enabled separation of the two products, to give (2S,3E)-**5b** (49%, 90.3% ee) and (2R,3E)-**5a** (40%, 97.0% ee), respectively (Scheme 2).<sup>14</sup> The enantiomeric ratio (*E* value) was 86.

Next, the epoxide (4), was treated with NaN<sub>3</sub> (5.0 equiv) and AcOH (2.5 equiv), α-azide ( $2R^*$ , $3R^*$ )-7 and β-azide ( $2R^*$ , $3R^*$ )-8 (quant. 6:1) formed, and pure  $7^{15}$  (81%) was accessible by recrystallization of the crude product. α-Substitution was confirmed with an authentic product<sup>16</sup> derived from a cyclic sulfate 9. If this epoxide ring-opening reaction was performed with NH<sub>4</sub>Cl instead of AcOH, an undesired α-epimer ( $2R^*$ , $3S^*$ )-7 was accompanied (10%). Further transformation of pure 7 yielded the ( $\pm$ )-erythro-dihydrosphingosine ( $2R^*$ , $3S^*$ )- $2^{17}$  (84%) in Scheme 3.

OH
$$C_{14}H_{29}$$
 $CO_{2}H$ 
 $CO_{2}H$ 

**Scheme 2.** Reagents and conditions: (a) Amano PS, vinyl acetate, THF, 30 °C; (b)  $CH_2N_2/Et_2O$ ,  $CH_2Cl_2$ , 49% for (2S,3E)-**5b**, 40% for (2R,3E)-**5a**.

Scheme 3. Reagents and conditions: (a) NaN<sub>3</sub>, AcOH, DMF, 50 °C, then recrystallization, 81% for (2*R*\*,3*R*\*)-7; (b) LiBH<sub>4</sub>, THF, rt, 92%; (c) H<sub>2</sub>, Pd–C, EtOH–AcOEt, rt, 90%.

At the stage of the two segments in hand, an unprecedented direct enzyme-mediated coupling in preparative scale was tried. *C. antarctica* lipase<sup>18</sup>-catalyzed acylation of  $(\pm)$ -2 with (2R,3E)-5a (1.5 equiv) worked under the conditions at high temperature  $(65 \, ^{\circ}\text{C})^{19}$  and reduced pressure  $(80 \, ^{\circ}\text{C})^{19}$  to ensure high specific activity of enzyme, and to remove concomitantly formed MeOH.

A tertiary alcohol,<sup>21</sup> 3-methyl-3-pentanol, was effective solvent, which has no effect on the enzyme-catalyzed reaction. The reaction proceeded in non-enantioselective manner, as both enantiomers of **2** were entirely consumed. Acetylation of the crude product afforded

OH OH 
$$C_{15}H_{31}$$
 OH  $C_{15}H_{31}$  OH  $C_{15}H_{31}$  OH  $C_{15}H_{31}$  OH  $C_{15}H_{31}$  OH  $C_{15}H_{31}$  OR  $C_{15}H_{31}$  OR  $C_{15}H_{31}$  OR  $C_{14}H_{29}$  OR  $C_{15}H_{31}$  OR  $C_{1$ 

**Scheme 4.** Reagents and conditions: (a) (2R,3E)-5a (1.5 equiv), Chirazyme L-2,  $(CH_3CH_2)_2C(OH)CH_3$ , 65 °C, 80 Torr; (b)  $Ac_2O$ , pyridine; (c)  $Et_3N$ , MeOH, 38% for (2S,3R,2'R)-1a, 37% for (2R,3S,2'R)-1a.

natural (2S,3R,2'R)-**1b** and its (2R,3S,2'R)-isomer, (total 75%), which were well separated on chromatography. Each was deacetyated (quant.) to give symbioramide (**1a**) and the unnatural diastereomer (Scheme 4). Their relative configurations were undoubtedly confirmed as tris *p*-bromobenzoates (2S,3R,2'R)-**1c** and (2R,3S,2'R)-**1c**, respectively.<sup>22</sup>

In conclusion, natural (2S,3R,2'R)-symbioramide (1a) and (2R,3S,2'R)-isomer was synthesized by lipase-catalyzed coupling between  $(\pm)$ -dihydrosphingosine (2) and (R)- $\beta$ , $\gamma$ -unsaturated  $\alpha$ -hydroxyester 5a, both of which were derived from the common intermediate, methyl  $(\pm)$ -trans-2,3-epoxyoctadecanonate (4). The number of steps involved in all of the convergent routes totaled only 12.

## Acknowledgements

We deeply thank Professor emeritus Kenji Mori, the University of Tokyo, for information and spectra on synthetic symbioramide and its diastereomer, and Amano Enzyme Inc. for lipase PS. This work was accomplished as the 21st Century COE Program (KEIO LCC), also supported by 'SORST: Solution-Oriented Research for Science and Technology' of Japan Science and Technology Corporation, and we express our sincere thanks to Professors Keisuke Suzuki, Takashi Matsumoto and Dr. Ken Ohmori of Tokyo Institute of Technology.

## References and notes

- Kobayashi, J.; Ishibashi, M.; Nakamura, H.; Hirata, Y.; Yamasu, T.; Sasaki, T.; Ohizumi, Y. Experientia 1988, 44, 800–802.
- Yoshida, J.; Nakagawa, M.; Seki, H.; Hino, T. J. Chem. Soc., Perkin 1 1992, 343–350.
- 3. Mori, K.; Uenishi, K. Liebigs Ann. Chem. 1994, 41-48.
- Azuma, H.; Takao, R.; Niiro, H.; Shikata, K.; Tamagaki, S.; Tachibana, T.; Ogino, K. J. Org. Chem. 2003, 68, 2790–2797.
- 5. Legters, J.; Thijs, L.; Zwanemburg, B. Recl. Trav. Chim. Pays-Bas 1992, 111, 1-15, This step became highly reproducible as follows. To a soln of methyl (E)-2octadecenoate (10.8 g, 36.4 mmol) in dry Cl(CH<sub>2</sub>)<sub>2</sub>Cl (150 mL) was added m-CPBA (9.5 g, 55.1 mmol) and 4,4-thiobis-(6-*tert*-butyl-3-methylphenol) (950 mg, mmol) at 0 °C. The reaction mixture was degassed to remove O<sub>2</sub> by applying ultrasonic (70–80 W) irradiation under evacuation at rt and was stirred under reflux for 2.5 h. The reaction was quenched by  $Na_2S_2O_3$  soln (100 mL) and the mixture was stirred at rt for another 20 min. The mixture was filtered through Celite and filtrate was concentrated in vacuo. The residue was purified by  $SiO_2$  (400 g, hexane/AcOEt = 12:1), and recrystallization from hexane gave 4 (10.6 g, 94%) as colorless fine needles, mp 44.0–44.2 °C; <sup>1</sup>H NMR  $\delta$  0.86 (t, 3H, J 6.8 Hz), 1.23 (m, 24H), 1.42–1.48 (m, 2H), 1.56–1.63 (m, 2H), 3.13 (ddd, 1H, J 2.0, 5.9, 6.3 Hz), 3.20 (d, 1H, J 2.0 Hz), 3.75 (s, 3H).
- 6. (a) Hartman, B. C.; Rickborn, B. J. Org. Chem. 1972, 37, 943–946; (b) Inokuchi, T.; Kusumoto, M.; Matsumoto, S.;

- Okada, H.; Torii, S. *Chem. Lett.* **1991**, 2009–2012; (c) Tomitani, K.; Sato, K.; Takahashi, H. *Jpn. Kokai Tokkyo Koho* **2001**, JP 2001039973 (CA 134:162911).
- Mg(ClO<sub>4</sub>)<sub>2</sub> as a Lewis acid promoter for t-Bu ether synthesis: Bartoli, G.; Bosco, M.; Locatelli, M.; Marcantoni, E.; Melchiorre, P.; Sambri, L. Org. Lett. 2005, 7, 427–430.
- 8. To a mixture of Mg(ClO<sub>4</sub>)<sub>2</sub> (9.1 g, 30.6 mmol), MS3A (13.9 g) in dry toluene (150 mL) was added a soln of **4** (8.7 g, 27.8 mmol) in toluene (50 mL). The mixture was stirred under reflux for 2 h and then quenched by water (100 mL). Conventional workup and purification by SiO<sub>2</sub> (400 g, hexane/AcOEt = 10:1) to give **5a** (5.5 g, 61%, E/Z = 4:1) as colorless oil. The E/Z ratio was estimated by comparing <sup>1</sup>H NMR: (E)  $\delta$  5.47 (dd); (Z)  $\delta$  5.35–5.41 (m).
- 9. Mp 54.0–55.5 °C; IR 2918, 2852, 1741, 1703, 1628, 1471, 1275, 1146, 1093, 982, 717 cm $^{-1}$ ;  $^{1}$ H NMR (CDCl<sub>3</sub>)  $\delta$  0.85 (t, 3H, J 6.9 Hz), 1.23–1.39 (m, 24H), 1.43–1.49 (m, 2H), 2.28 (ddt, 2H, J 1.5, 6.8, 6.8 Hz), 3.86 (s, 3H), 6.63 (dt, 1H, J 15.9, 1.5 Hz), 7.18 (dt, 1H, J 15.9, 6.8 Hz);  $^{13}$ C NMR (CDCl<sub>3</sub>)  $\delta$  14.21, 22.75, 27.78, 29.25, 29.41, 29.54, 29.66, 29.70, 29.71 (×3), 29.74, 32.00, 33.22, 52.90, 124.86, 155.50, 162.61, 182.81; Anal. Calcd for  $C_{19}H_{36}O_{3}$ : C, 73.03; H, 11.61. Found: C, 73.15; H, 11.52.
- Sugimura, H.; Yoshida, K. J. Org. Chem. 1993, 58, 4484–4486.
- 11. <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.88 (t, 3H, *J* 6.8 Hz), 1.23 (br s, 24H), 2.03 (dt, 2H, *J* 7.0, 7.0 Hz), 2.87 (br s, 1H), 3.77 (s, 3H), 4.58 (d, 1H, *J* 5.9 Hz), 5.47 (dd, 1H, *J* 5.9, 15.1 Hz), 5.86 (dt, 1H, *J* 15.1, 7.0 Hz).
- 12. A mixture of **5a** (E/Z = 22:1, 3.1 g, 9.9 mmol), phosphate buffer (pH 7.0, 0.1 M, 100 mL) and Chirazyme L-2 (c-f, 680 mg) was stirred vigorously at 30 °C for 34 h. The mixture was diluted with water, and pH was adjusted to 4 by 1 M HCl, and mixed with AcOEt. The conventional workup gave the product (3.1 g, quant.) as white solid, and recrystallization from acetone gave 3a (2.9 g, 98%) as colorless fine needles, mp 96.0-96.2 °C; IR 3450, 2916, 2850, 1747, 1471, 1246, 1105, 1061, 964, 889, 652 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.81 (t, 3H, J 6.8 Hz), 1.16–1.25 (m, 24H), 1.29–1.39 (m, 2H), 2.00 (dt, 2H, J 7.1, 7.1 Hz), 4.62 (d, 1H, J 6.4 Hz), 5.49 (dd, 1H, J 15.3, 6.4 Hz), 5.87 (dt, 1H, J 15.3, 7.1 Hz).  $^{13}$ C NMR (DMSO- $d_6$ )  $\delta$  13.96, 22.14, 28.59, 28.77, 28.91, 29.10 (×6), 31.35, 31.54, 70.78, 128.02, 131.87, 174.28; Anal. Calcd for C<sub>18</sub>H<sub>34</sub>O<sub>3</sub>: C, 72.44; H, 11.48. Found: C, 72.52; H, 11.39. The hydrolysis proceeded in non-enantioselective manner. Also the epoxy ester (±)-4 was a good substrate for Chirazyme L-2, to give the epoxy acid (quant.).
- 13. Sugai, T.; Ohta, H. *Tetrahedron Lett.* **1991**, *32*, 7063–7064, Long chain α-hydroxy acid is superior to methyl and benzyl esters to acquire higher enantioselectivity in resolution.
- 14. To a soln of **3a** (625 mg, 2.1 mmol) in THF (7 mL) and vinyl acetate (7 mL) was Amano PS (2.0 g), and the mixture was stirred at 30 °C for 4 days. The mixture was filtered and the filtrate was concentrated in vacuo. The residue was diluted with CH<sub>2</sub>Cl<sub>2</sub> (2 mL), and was added CH<sub>2</sub>N<sub>2</sub>/Et<sub>2</sub>O to give a mixture of (2*R*,3*E*)-**5a** and (2*S*,3*E*)-**5b**, which was purified by SiO<sub>2</sub> (100 g, hexane/ AcOEt = 8:1) to give (2*R*,3*E*)-**5a** (260 mg, 40%, 97% ee) and (2*S*,3*E*)-**5b** (363 mg, 49%, 90% ee) as colorless oil. (2*R*,3*E*)-**5a**: [ $\alpha$ ]<sub>D</sub><sup>20</sup> -42.4 (*c* 0.50, CHCl<sub>3</sub>) [Ref. 2 [ $\alpha$ ]<sub>D</sub><sup>19</sup> -44.7 (*c* 0.257, CHCl<sub>3</sub>), Ref. 10 [ $\alpha$ ]<sub>D</sub><sup>20</sup> -45.3 (*c* 0.70, CHCl<sub>3</sub>)]. (2*S*, 3*E*)-**5b**: [ $\alpha$ ]<sub>D</sub><sup>20</sup> +48.8 (*c* 1.17, CHCl<sub>3</sub>); IR 2924, 2854, 1751, 1437, 1373, 1228, 1049, 968 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.85 (t, 3H, *J* 6.8 Hz), 1.23 (m, 22H), 1.32–1.40 (m, 2H), 2.05 (dt, 2H, *J* 7.5, 7.5 Hz), 2.12 (s, 3H), 3.72 (s, 3H), 5.35

- (d, 1H, J 7.5 Hz), 5.50 (dd, 1H, J 15.4, 7.5 Hz), 5.91 (dt, 1H, J 15.4, 7.5 Hz);  $^{13}$ C NMR (CDCl<sub>3</sub>)  $\delta$  14.08, 20.68, 22.66, 28.52, 29.08, 29.33, 29.38, 29.54, 29.61 (×2), 29.64, 31.88, 32.26, 52.41, 73.30, 121.53, 138.26, 169.58, 170.14; Anal. Calcd for C<sub>19</sub>H<sub>38</sub>O<sub>4</sub>: C, 71.14; H, 10.80. Found: C, 70.95; H, 10.69. The ees were estimated by the  $^{1}$ H NMR spectra of MTPA esters **5c**. (2*R*,2'*S*)-**5c**: (CDCl<sub>3</sub>)  $\delta$  2.06 (dt, 1H, J 6.8, 6.8 Hz, H-5), 3.55 (s, 3H, MTPA-OCH<sub>3</sub>), 3.73 (s, 3H, CO<sub>2</sub>CH<sub>3</sub>), 5.58 (m, 2H, H-2, H-3), 5.97 (m, 1H, H-4); (2*S*,2'*S*)-**5c**:  $\delta$  2.02 (dt, 1H, J 7.0, 7.0 Hz, H-5), 3.64 (s, 3H, MTPA-OCH<sub>3</sub>), 3.77 (s, 3H, CO<sub>2</sub>CH<sub>3</sub>), 5.51 (d, 1H, J 15.1, 7.3 Hz, H-3), 5.58 (d, 1H, J 7.3 Hz, H-2), 5.86 (dt, 1H, J 15.1, 7.0 Hz, H-4).
- 15. A mixture of 4 (1.0 g, 3.2 mmol), NaN<sub>3</sub> (1.0 g, 16.0 mmol), AcOH (640 µL, 11.2 mmol) in dry DMF (31 mL) was stirred at 50 °C for 24 h. Water (30 mL) was added and extracted with Et<sub>2</sub>O. The conventional workup, chromatography (SiO<sub>2</sub>, 100 g, hexane/AcOEt = 8:1), and recrystallization from hexane to give pure  $(2R^*,3R^*)$ -7 (920 mg, 81%) as colorless needles, mp 41.0-41.5 °C; IR 3459, 2920, 2850, 2111, 1743, 1468, 1438, 1207, 721 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.86 (t, 3H, J 6.8 Hz), 1.23 (br s, 26H), 1.50 (m, 2H), 2.24 (d, 1H, J 5.4 Hz), 3.81 (s, 3H), 3.91 (m, 1H) 3.95 (d, 1H, J 5.9 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  14.07, 22.66, 25.34, 29.33, 29.46, 29.53, 29.63, 29.66 (×6), 31.89, 33.00, 52.71, 66.25, 71.92, 169.46; Anal. Calcd for C<sub>19</sub>H<sub>37</sub>N<sub>3</sub>O<sub>3</sub>: C, 64.19; H, 10.49; N, 11.82. Found: C, 64.22; H, 10.43; N, 11.73.  $(2R^*,3R^*)$ -8: <sup>1</sup>H NMR  $\delta$  3.09 (d, 1H, J 5.3 Hz), 3.50 (dt, 1H, J 10.3, 3.4 Hz), 4.31 (dd, J 5.3, 3.4 Hz); (2R\*,3S\*)-7:  $^{1}$ H NMR  $\delta$  2.00 (d, 1H, J 6.8 Hz), 4.04 (m, 1H);  $^{13}$ C NMR  $\delta$  52.82, 65.92, 72.31, 169.74.
- He, L.; Byun, H.-S.; Bittman, R. J. Org. Chem. 2000, 65, 7618–7626.
- 17. 10: colorless fine needles, mp 48.0-50.0 °C; IR 3357, 2919, 2850, 2103, 1467, 1265, 1049, 759 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.86 (t, 3H, J 6.8 Hz), 1.23 (br s, 26H), 1.49–1.53 (m, 2H), 2.25 (m, 2H), 3.40 (ddd, 1H, J 4.9, 4.9, 4.9 Hz), 3.75 (m, 1H), 3.87 (dd, 1H, J 1.5, 4.9 Hz), 3.88 (dd, 1H, J 1.5, 4.9 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 14.09, 22.66, 25.57, 29.33, 29.46, 29.51, 29.54, 29.63, 29.66 (×5), 31.89, 33.70, 62.45, 66.78, 72.58; Anal. Calcd for C<sub>18</sub>H<sub>37</sub>N<sub>3</sub>O<sub>2</sub>: C, 66.01; H, 11.39; N, 12.83. Found: C, 65.77; H, 11.31; N, 12.61. (2R\*,3S\*)-2: mp 62.5-63.0 °C; IR 3354, 3256, 2920, 2848, 1468, 1261, 1020, 802, 723 cm<sup>-1</sup>; <sup>13</sup>C NMR (pyridine- $d_5$ )  $\delta$ 14.26, 22.93, 26.63, 29.60, 29.91, 29.98 (×5), 30.08, 30.22, 32.11, 34.26, 35.09, 58.47, 64.63, 73.82; Anal. Calcd for C<sub>18</sub>H<sub>39</sub>NO<sub>2</sub>: C, 71.70; H, 13.04; N, 4.65. Found: C, 71.52; H, 13.11; N, 4.54. Its NMR spectra were identical with those reported previously: So, R. C.; Ndonye, R.; Izmirian, D. P.; Richardson, S. K.; Guerrera, R. L.; Howell, A. R. J. Org. Chem. 2004, 69, 3233–3235.
- Lipase-catalyzed aminolysis in organic solvent, review: Alfonso, I.; Gotor, V. Chem. Soc. Rev. 2004, 33, 201–209.
- Kurokawa, M.; Sugai, T. Bull. Chem. Soc. Jpn. 2004, 77, 1021–1025.
- Sugai, T.; Takizawa, M.; Bakke, M.; Ohtsuka, Y.; Ohta, H. *Biosci. Biotechnol. Biochem.* **1996**, *60*, 2059–2063.
- Kitaguchi, H.; Klibanov, A. M. J. Am. Chem. Soc. 1989, 111, 9272–9273.
- 22. A soln of **2** (60.2 mg, 0.20 mmol), (2R,3E)-**5a** (93.4 mg, 0.30 mmol), and 3-methyl-3-pentanol (3.5 mL) was added Chirazyme L-2 (250 mg) and stirred at 65 °C under 80 Torr for 4 h. The reaction mixture was concentrated in vacuo, and acetylated in the usual manner. After workup, the residue was purified by preparative TLC (hexane/AcOEt = 2:1) to give (2S,3R,2'R)-**1b** (53.4 mg, 38%) and (2R,3S,2'R)-**1b** (52.1 mg, 37%). (2S,3R,2'R)-**1b**: prisms from hexane–AcOEt,  $R_f$  0.43 [hexane/AcOEt (2:1)];  $[\alpha]_D^{20}$  -7.2 (c 0.58, CHCl<sub>3</sub>) [Ref. 3  $[\alpha]_D^{21}$  -5.09 (c 1.36,

CHCl<sub>3</sub>)]; mp 80.5–81.0 °C (Refs. 1 and 2 75–78 °C, Ref. 3 78–79 °C); ÎR 3313, 2850, 1736, 1664, 1541, 1468, 1385, 1269, 1039, 960 cm<sup>-1</sup>; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.86 (t, 3H), 1.23-1.39 (m, 50H), 2.02 (s, 3H, Ac), 2.05 (s, 3H, Ac), 2.16 (s, 3H, Ac), 4.01 (dd, 1H, J 2.9, 11.2 Hz, H-1), 4.28 (dd, 1H, J 6.8, 11.2 Hz, H-1), 4.32–4.36 (m, 1H, H-2), 4.88 (dt, 1H, J 7.8, 5.1 Hz, H-3), 5.46 (d, 1H, J 7.3 Hz, H-2'), 5.49 (dd, 1H, J 7.3, 13.8 Hz, H-3'), 5.88 (dt, 1H, J 13.8, 6.8 Hz, H-4'), 6.51 (d, 1H, J 8.3 Hz, NH);  $^{13}$ C NMR (CDCl<sub>3</sub>)  $\delta$ 14.11, 20.75, 20.95, 22.67, 25.32, 28.66, 29.22, 29.35, 29.45, 29.56, 29.59, 29.64, 29.69 (×17), 31.40, 31.91, 32.29, 50.68, 62.14, 73.81, 74.50, 122.80, 137.84, 168.53, 169.38, 170.86, 171.27. Anal. Calcd for C<sub>42</sub>H<sub>77</sub>NO<sub>7</sub>: C, 71.24; H, 10.96; N, 1.98. Found: C, 71.24; H, 10.88; N, 1.93. Its <sup>1</sup>H NMR spectrum was identical with those reported previously. 1-3 (2R,3S,2'R)-1b: fine needles from hexane–AcOEt,  $R_f$  0.33;  $[\alpha]_{D}^{20}$  -28.3 (c 0.61, CHCl<sub>3</sub>); mp 82.5–83.0 °C; IR 3323, 2848, 1747, 1664, 1548, 1465, 1373, 1226, 1041, 976 cm <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.86 (t, 6H, J 6.8 Hz), 1.23–1.36 (m, 50H), 2.03 (s, 3H, Ac), 2.05 (s, 3H, Ac), 2.15 (s, 3H, Ac), 4.08 (dd, 1H, J 4.2, 11.6 Hz, H-1), 4.19 (dd, 1H, J 6.8, 11.6 Hz, H-1), 4.36 (m, 1H, H-2), 4.82 (dt, 1H, J 8.3, 4.3 Hz, H-3), 5.45 (d, 1H, J 7.3 Hz, H-2'), 5.49 (dd, 1H, J 7.3, 14.4 Hz, H-3'), 5.88 (dt, 1H, J 14.4, 7.0 Hz, H-4'), 6.52 (d, 1H, J 8.8 Hz, NH); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 14.11, 20.77, 20.91, 22.67, 25.37, 28.64, 29.20, 29.31, 29.35, 29.45, 29.53, 29.58, 29.68 (×17), 31.57, 31.91, 32.31, 50.52, 62.22, 74.42, 74.55, 122.86, 138.03, 168.46, 169.40, 171.03, 171.11. Anal. Calcd for C<sub>42</sub>H<sub>77</sub>NO<sub>7</sub>: C, 71.24; H, 10.96; N, 1.98. Found: C, 71.16; H, 10.91; N, 1.92. Each was treated with Et<sub>3</sub>N–MeOH for deacetylation. (2*S*,3*R*,2'*R*)-1a: white solid,  $R_{\rm f}$  0.34 [CHCl<sub>3</sub>/MeOH (10:1)];  $[\alpha]_{\rm D}^{19}$  +5.3 (*c* 0.32, CHCl<sub>3</sub>) [Ref. 1  $[\alpha]_{\rm D}^{22}$  +5.8 (*c* 1, CHCl<sub>3</sub>), Ref. 3  $[\alpha]_{\rm D}^{19}$ +3.6 (c 0.31, CHCl<sub>3</sub>)]; mp 112–113 °C (Ref. 1 105–107 °C, Ref. 3 115.0–116.5 °C); <sup>1</sup>H NMR  $\delta$  0.87 (t, 6H, J 6.8 Hz), 1.23 (br s, 50H), 1.36 (m, 2H), 2.07 (dt, 2H, J 7.2, 7.2 Hz, H-5'), 2.38 (m, 2H, OH), 2.98 (br s, 1H, OH), 3.74–3.83 (m, 3H, H-1, H-2, H-3), 4.00 (dd, 1H, J 3.4, 11.2 Hz, H-1), 4.51 (d, 1H, J 7.2 Hz, H-2'), 5.56 (dd, 1H, J 7.2, 15.5, Hz, H-3'), 5.89 (dt, 1H, J 15.5, 7.2 Hz, H-4'), 6.94 (d, 1H, J 7.2 Hz, NH); HRMS (FAB<sup>+</sup>): calcd for  $C_{36}H_{71}O_4NNa$ :  $(M+Na^+)$ : 604.5281; found: m/z 604.5308. Its NMR spectrum was identical with those reported previously. 1-4 (2R,3S,2'R)-1a: white solid,  $R_f$  0.43;  $[\alpha]_D^{20}$  -36.1 (c 0.15, CHCl<sub>3</sub>); [Ref. 3 [ $\alpha$ ]<sub>D</sub><sup>18</sup> +42.9 (c 1.5, CHCl<sub>3</sub>) for (2S,3R,2'S)-1a]; mp 108–109 °C; [Ref. 3 99.5–100.5 °C for (2S,3R,2'S)-1a]; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.87 (t, 6H, J 6.8 Hz), 1.25 (br s, 50H), 1.48 (m, 2H), 2.08 (dt, 2H, J7.2, 7.2 Hz, H-5'), 2.33 (m, 1H, OH), 2.94 (br s, 1H, OH), 3.62 (m, 1H, OH), 3.72-3.83 (m, 3H, H-1, H-2, H-3), 4.01 (dd, 1H, J 3.2, 11.5 Hz, H-1), 4.51 (d, 1H, J 7.2 Hz, H-2'), 5.55 (dd, 1H, J 7.2, 15.1 Hz, H-3'), 5.89 (dt, 1H, J 15.1, 7.2 Hz, H-4'), 6.89 (d, 1H, J 7.2 Hz, NH). HRMS (FAB<sup>+</sup>): calcd for  $C_{36}H_{71}O_4NNa$ : [M+Na<sup>+</sup>]: 604.5281; found: 604.5264. Its NMR spectrum was identical with that reported previously.<sup>3</sup> (2S, 3R, 2'R)-1c: white solid,  $R_f$  0.48 [hexane/AcOEt (5:1)];  $^{1}$ H NMR (CDCl<sub>3</sub>)  $\delta$  0.87 (t, 6H, J6.8 Hz), 1.21–1.39 (m, 50H), 1.75 (m, 1H, H-4), 1.85 (m, 1H, H-4), 2.03 (dt, 2H, J 7.1,7.1 Hz, H-5'), 4.42 (dd, 1H, J 4.0, 11.7, Hz, H-1), 4.54 (dd, 1H, J 5.4, 11.7, Hz, H-1), 4.59 (m, 1H, H-2), 5.22 (dt, 1H, J 8.3, 4.3 Hz, H-3), 5.62 (dd, 1H, J 7.1, 15.3 Hz, H-3'), 5.73 (d, 1H, J 7.3 Hz), 5.96 (dt, 1H, J 15.3, 6.8 Hz, H-4'), 7.02 (d, 1H, J 8.3 Hz, NH), 7.37, 7.46, 7.59, 7.57, 7.68, 7.84 (each d, J 8.4 Hz, 12H, aromatic H). (2R,3S,2'R)-1c: white solid,  $R_f$  0.58; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.86 (t, 6H, J 6.8 Hz), 1.23–1.39 (m, 50H), 1.78 (m, 2H, H-4), 1.89 (dt, 2H, J 7.0, 7.0 Hz, H-5'), 4.46 (dd, 1H, J 4.2, 11.4 Hz, H-1), 4.56 (dd, 1H, J 7.3, 11.4 Hz, H-1), 4.65 (m, 1H, H-2), 5.17 (dt, 1H, *J* 7.8, 4.2 Hz, 3-H), 5.49 (dd, 1H, *J* 7.0, 15.3 Hz, H-3'), 5.73 (d, 1H, *J* 7.0 Hz, H-2'), 5.88 (dt, 1H, *J* 15.3, 7.0 Hz, H-4'), 6.88 (d, 1H, *J* 8.8 Hz, NH), 7.49 (m, 6H, aromatic

H), 7.66, 7.75, 7.84 (each d, J 8.5 Hz, 6H, aromatic H). NMR spectrum was identical with that for (2S,3R,2'S)-1c.<sup>3</sup>